

Research Overview of Clean Energy

Dr. Tae-Jin Kang

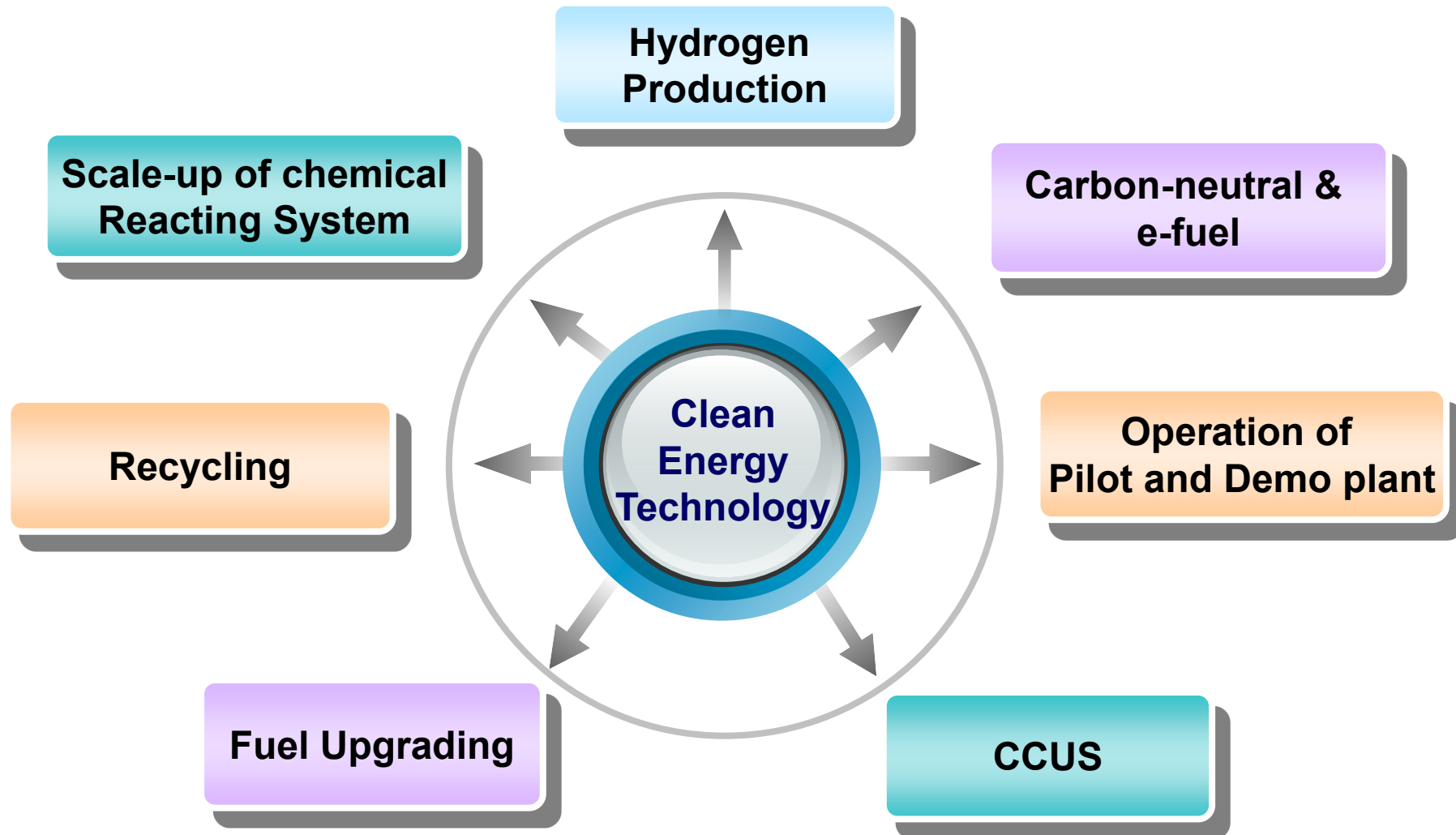
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Korea-EU Horizon Europe Researchers Networking Forum on Energy

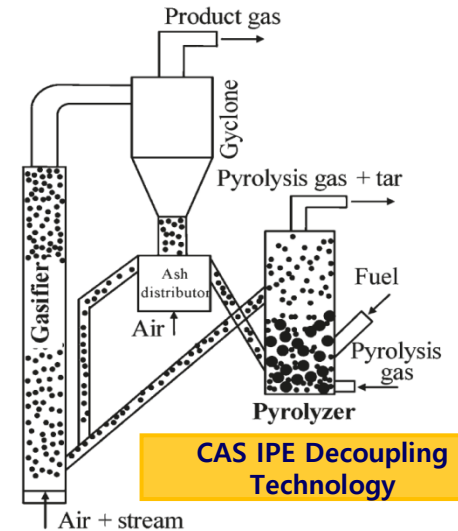
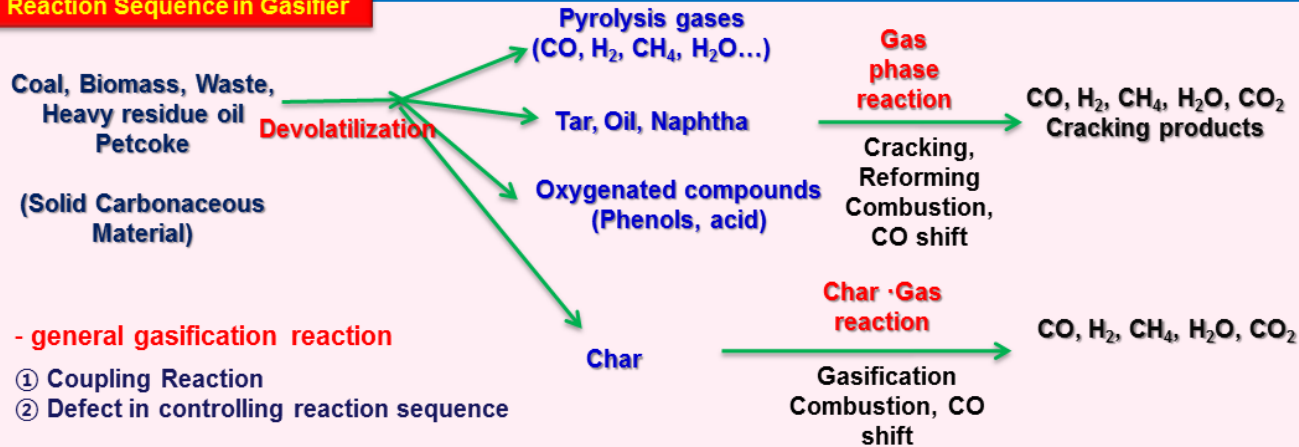


Research Accomplishments

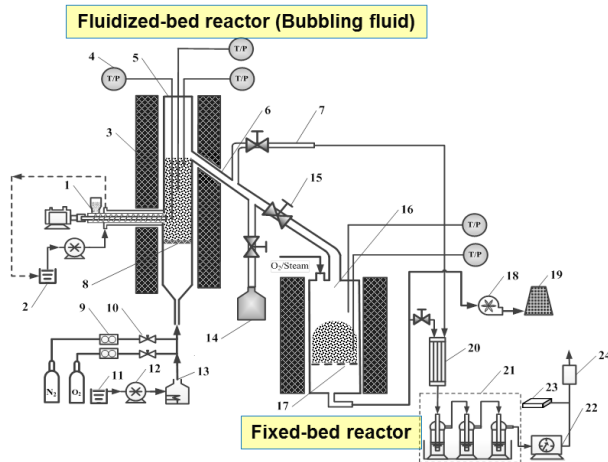


Decoupling Gasification

Reaction Sequence in Gasifier



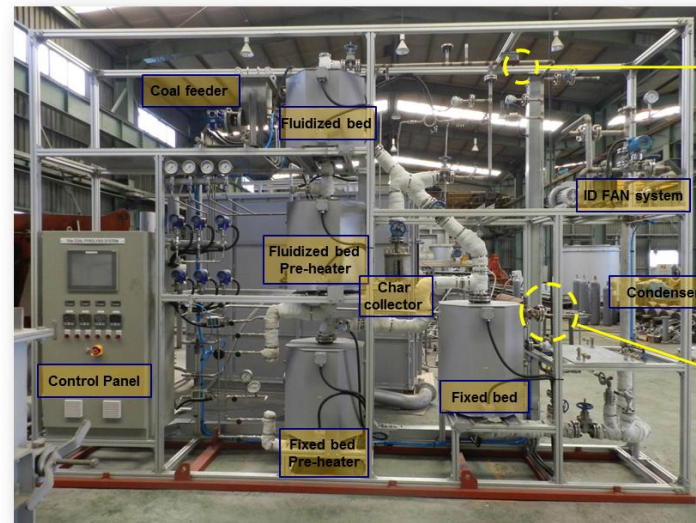
Developing hybrid FB process with CAS IPE



Schematic diagram of dual-bed coal gasification system

1-Screw feeder; 2-condensate pump; 3-electric furnace; 4-temperature and pressure detector; 5-fluidized bed reactor; 6-overflow tube; 7-gas and tar sampling pipe; 8-gas distribution plate; 9-mass flow meter; 10-needle valve; 11-water tank; 12-pump; 13-steam generator; 14-char collector; 15-ball valve; 16-tar reforming reactor; 17-char bed layer; 18-draught fan; 19-chimney; 20-condenser; 21-ice-water tank; 22-wet gas flow meter; 23-gas chromatograph; 24-pump

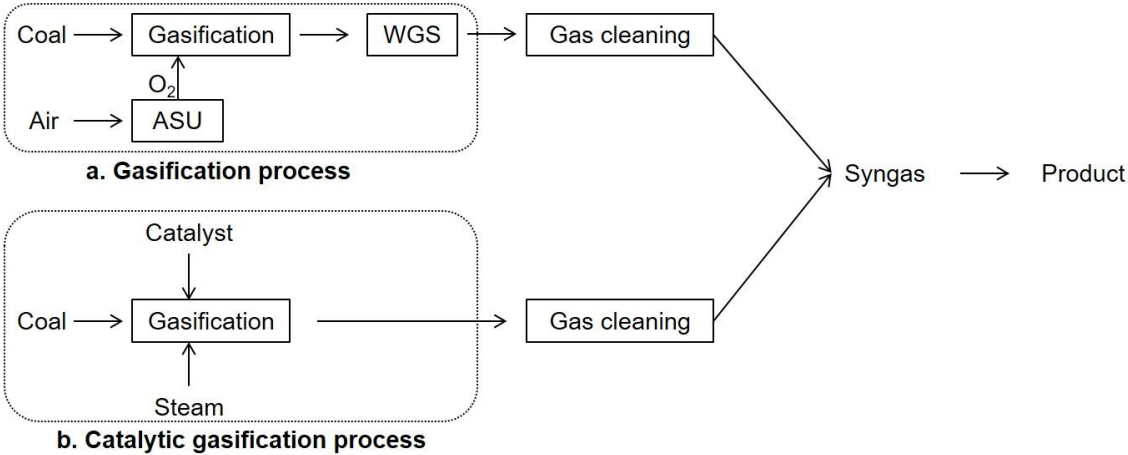
➢ 1kW Dual-bed gasification system (24 kg/day)



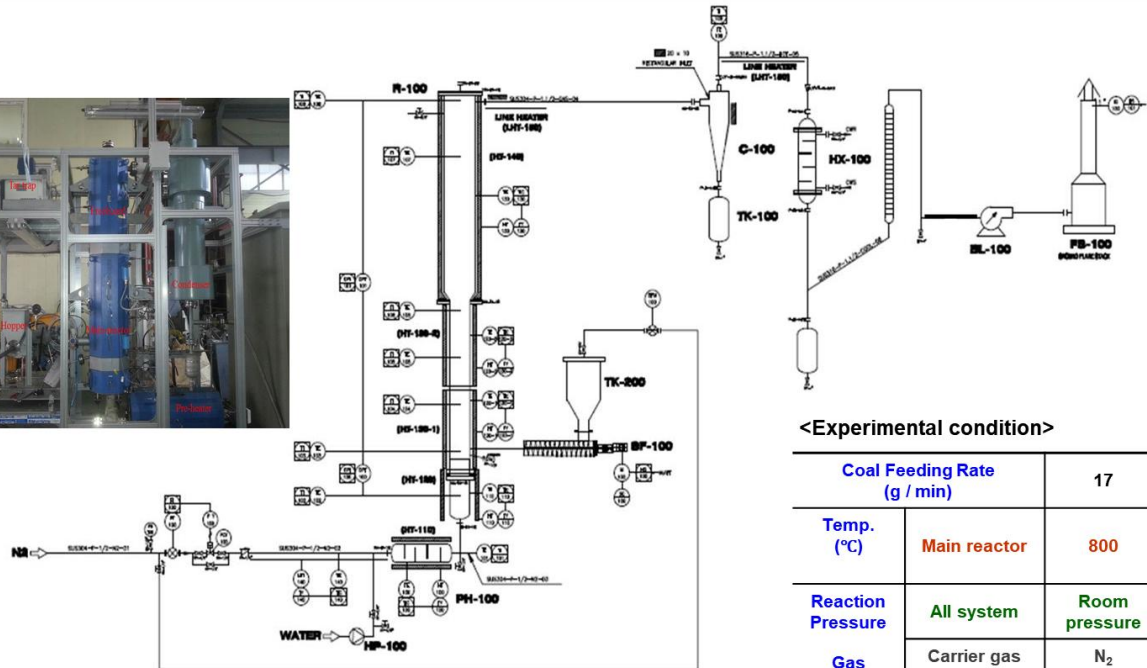
• Installation place : Dangjin in Korea



Steam-Catalytic Gasification



Bench-scale fluidized bed for steam-catalytic gasification



<Experimental condition>

Coal Feeding Rate (g / min)	17	
Temp. (°C)	Main reactor	800
Reaction Pressure	All system	Room pressure
Gas Stream	Carrier gas	N ₂
	Oxidant	Steam

Catalyst recovery system

Procedure of catalyst recovery

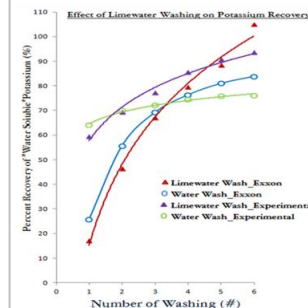
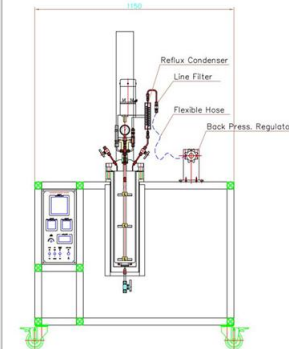
Impregnation coal → Catalytic gasification with steam → Syngas

Char

Catalyst recovery
(T: 150 °C, P: 20 bar, reactor volume: 5L)

1. Char + H₂O (H₂O: 2kg, N₂ condition) → water drain
2. Residue from 1 + limewater (H₂O: 2kg, CaO: 2.5g, CO₂ condition) → water drain

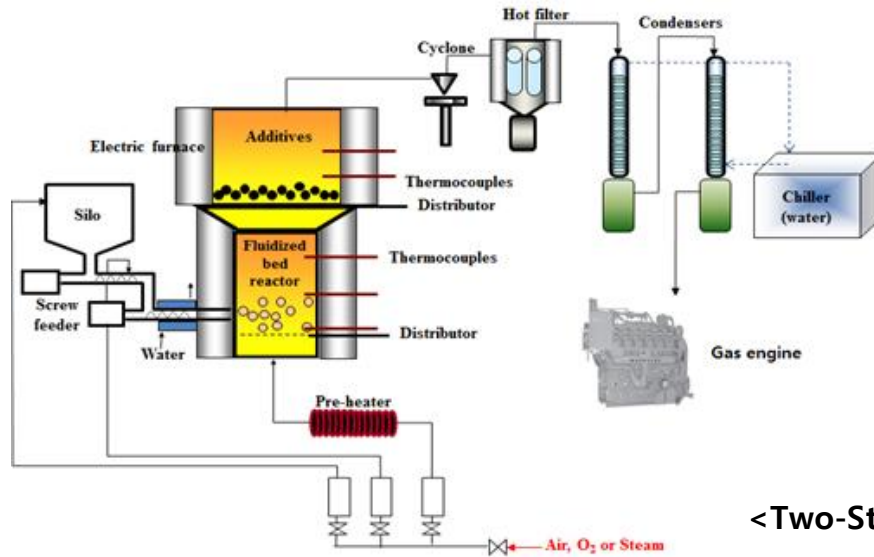
- The catalyst was recovered with a high yield (87%), by evaporating water according to methods 1 and 2.
- To recover potassium, the required energy required to boil 30°C water was 610 kcal/kg.



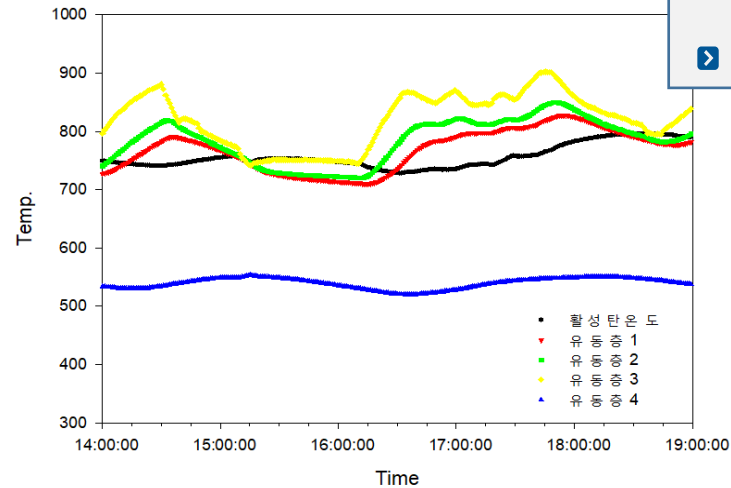
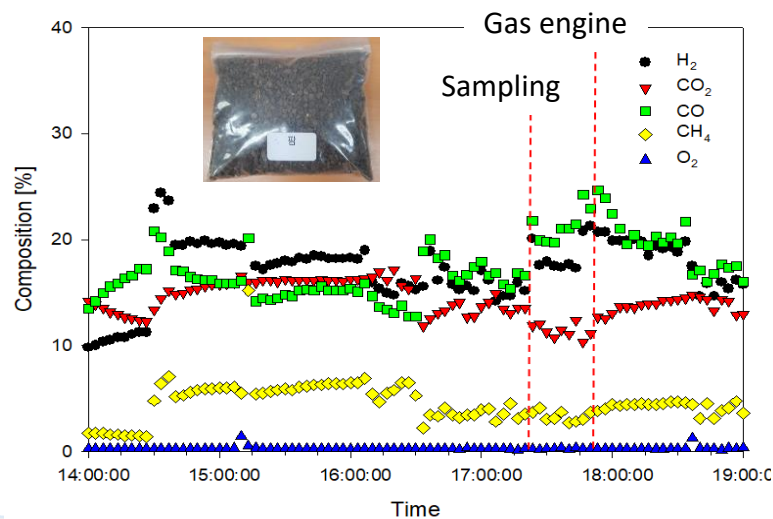
<Experimental Condition>

- FB Gasification**
- Catalytic steam gasification (at 800 °C, K₂CO₃)
 - Carbon conversion: 90%
 - Cold gas efficiency: 63%
- Catalyst Recovery**
- Catalyst recovery efficiency: 80~90%

Two-Stage Biomass Gasification



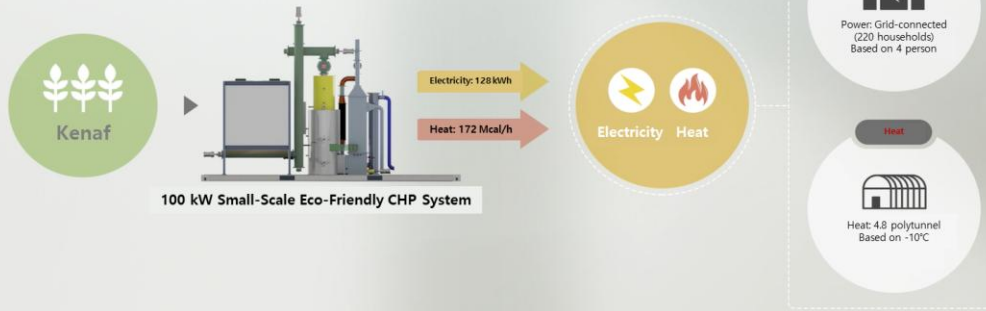
<Two-Stage Biomass Gasification & Gas Engine Power Generation >



- ▶ H₂ content: 26%
- ▶ Tar Removal: 99.8%
- ▶ Power Output: 33kW

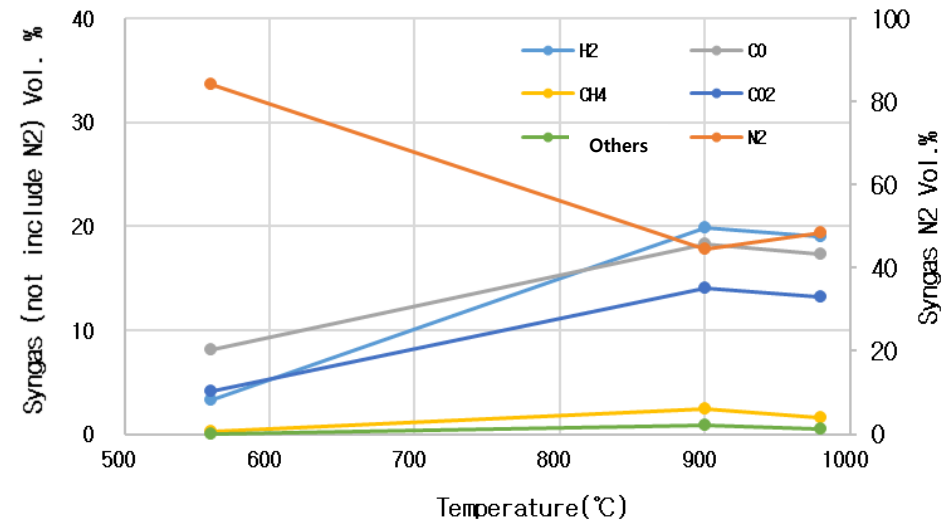
Herbaceous Biomass Gasification

Herbaceous Biomass Gasification



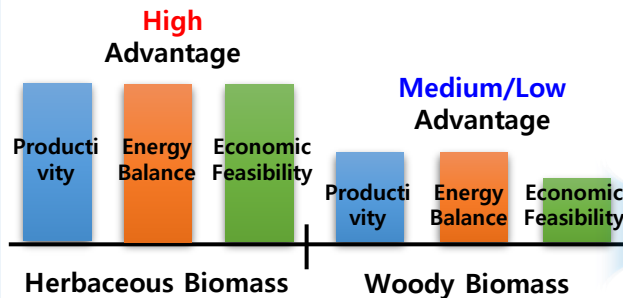
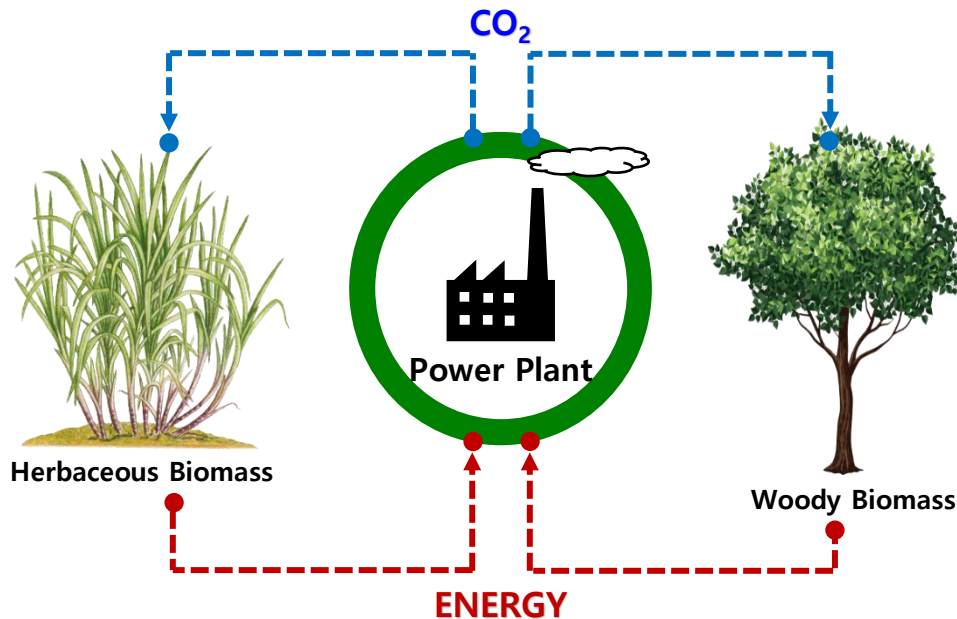
- Increasing temperature leads to higher yields of H_2 , CO , and CO_2 , while N_2 decrease.
- Near $980^\circ C$, the concentrations of H_2 , CO , and CO_2 slightly decrease, while N_2 increase.

Effect of Gasification Temperature on Syngas Composition

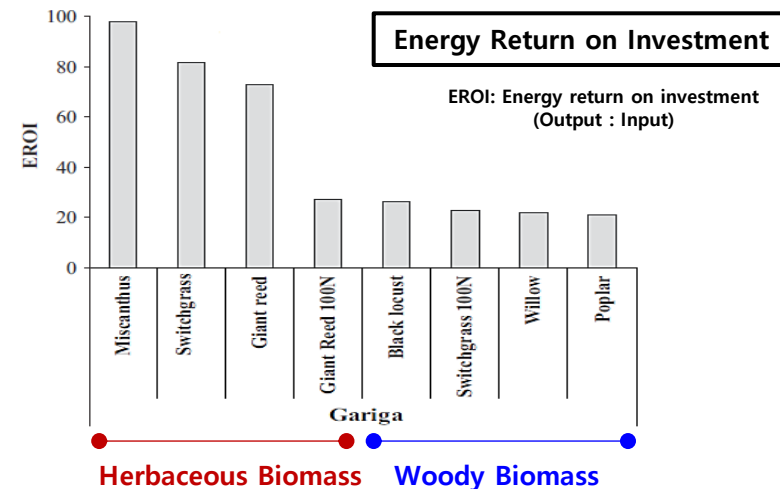
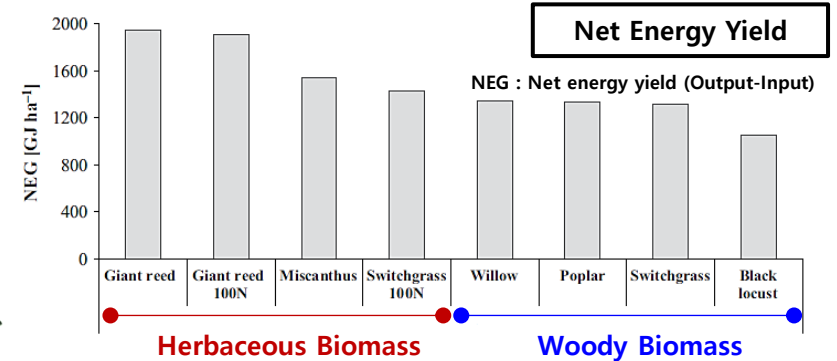


Advantages of Herbaceous Biomass

Which one is better as a CO₂ reduction fuel ?

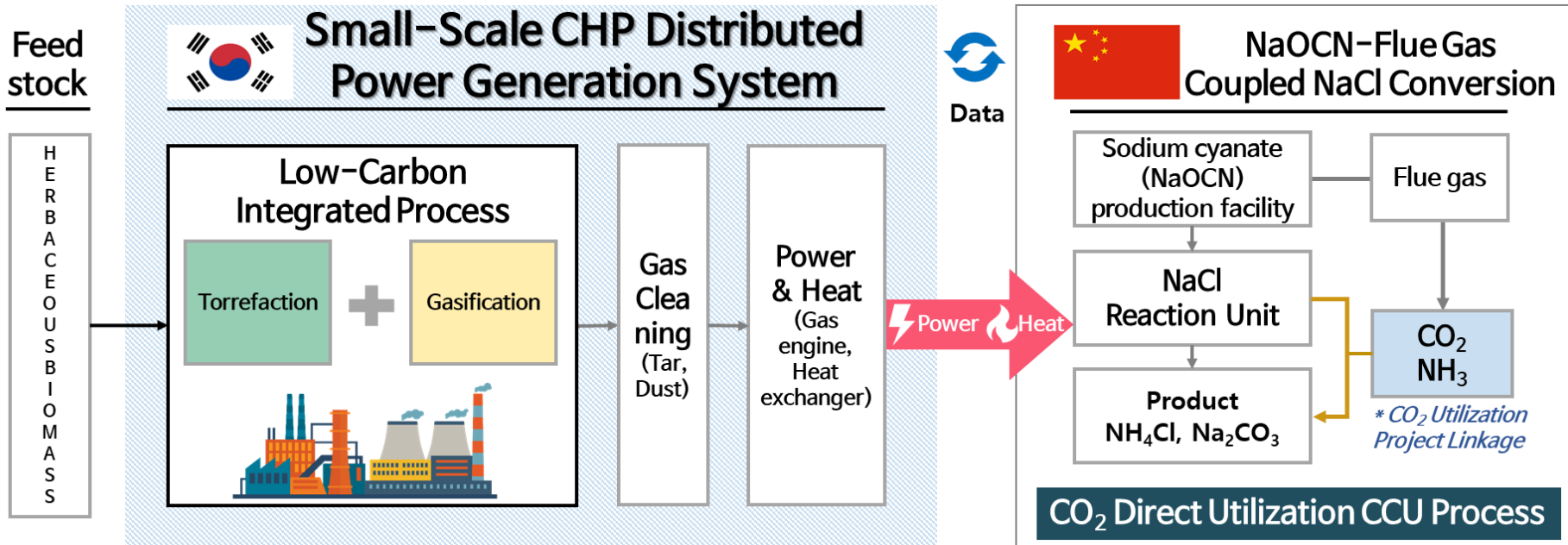


Ref.: S. Amaducci et al, GCB Bioenergy (2017) 9, 31-45.



Technology development for utilizing herbaceous biomass as an alternative fuel is essential for achieving carbon neutrality by 2050 and diversifying the energy mix

Distributed Herbaceous Biomass CHP integrated with CCU



▪ Direct CO_2 Utilization Technology Linked with Distributed Power Generation Using Herbaceous Biomass

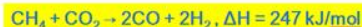
- ① Integrated torrefaction/gasification distributed power generation using herbaceous biomass
- ② Production of Na_2CO_3 and NH_4Cl utilizing NH_3 and CO_2 contained in waste gas from the NaOCN manufacturing process

Carbon-neutral process for reducing fine dust and greenhouse gas emissions



Dry Reforming of Methane (1)

Dry Reforming of Methane



- Produced synthesis gas (CO and H_2) can be used...
- Production of oxygenate compounds, acetic acid synthesis
- Chemical intermediates, hydrogenation of unsaturated compounds

Major problem is...

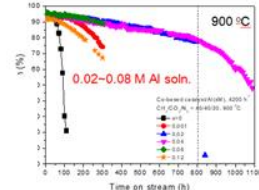
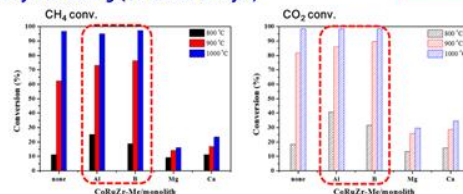
- Endothermic reaction → needs high reaction temperature
→ **Active metal aggregation**
- Carbon formation:** $\text{CH}_4 \rightarrow \text{C} + 2\text{H}_2$ (CH_4 decomposition)
or $2\text{CO} \rightarrow \text{CO}_2 + \text{C}$ (Boudouard reaction)

CH₄ decomposition, CO disproportionation and CH₄/CO₂ reaction



Lab-scale reforming reactor

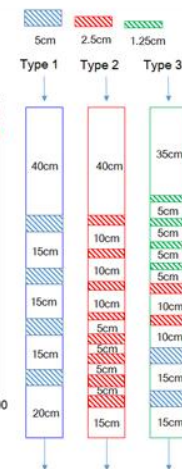
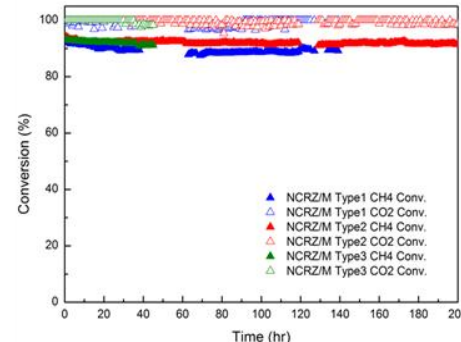
Catalyst screening (Co-based catalyst)



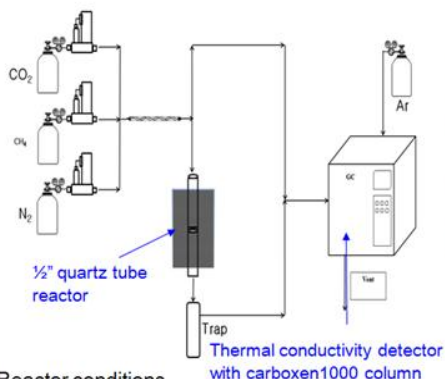
Addition of small amount of Al
→ enhanced catalyst stability
(10 times ↑)

- Pretreatment = 1000 °C, 10 h, feed cond.
- $\text{CH}_4/\text{CO}_2/\text{N}_2$ (%) = 40/40/20
- GHSV = 4,200 h^{-1} (109 $\text{cm}^3 \text{ min}^{-1}$)
- T = 800-1000 °C

Reaction condition
T = 1000 °C, P = 1 bar,
 CO_2/CH_4 = 50:50 (5.6 L/min) SV = 4,425 h^{-1}



Lab-scale reforming reactor



Reactor conditions

- ✓ Pressure: 0.1 Mpa
- ✓ Max temperature: 1,100 °C
- ✓ Mass flow rate (Brooks): 400 cm^3/min

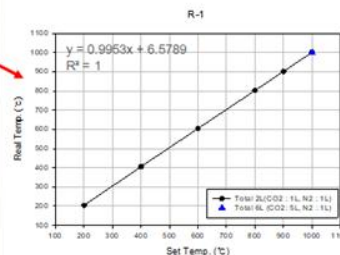


Bench-scale reforming reactor



Reactor size: 1 inch x 1.5 m
Material: alloy 600
Design Pressure: 3.5 barg
Design Temp.: 1500 °C

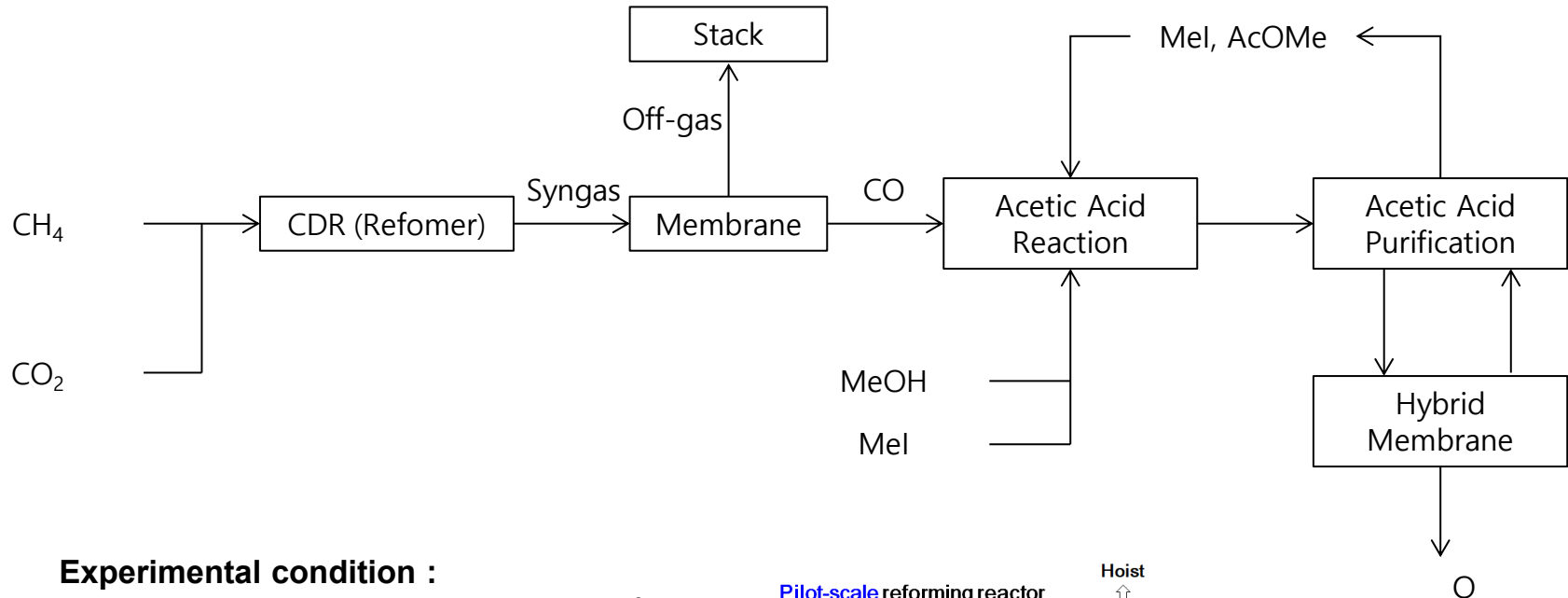
- Fixed-bed Inconel reactor
- ID: 2.29 cm, H: 150 cm



Reforming catalyst:
production of 2.8
kg CO per day

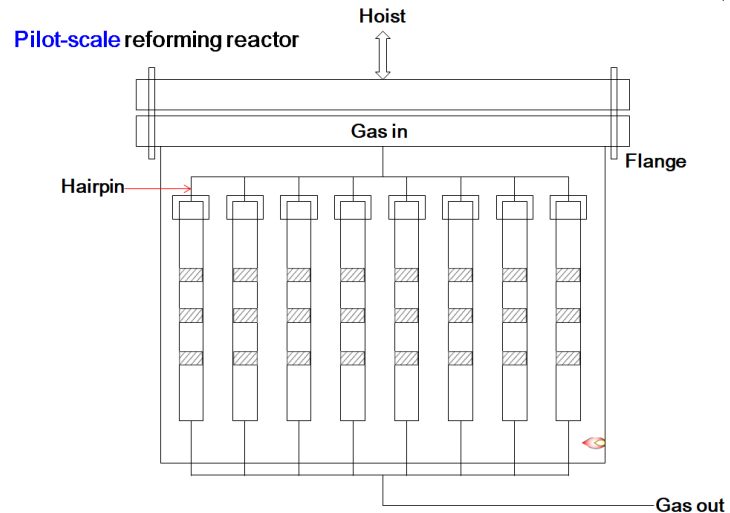
Optimal catalyst bed
: 70~80 cm

Dry Reforming of Methane (2)

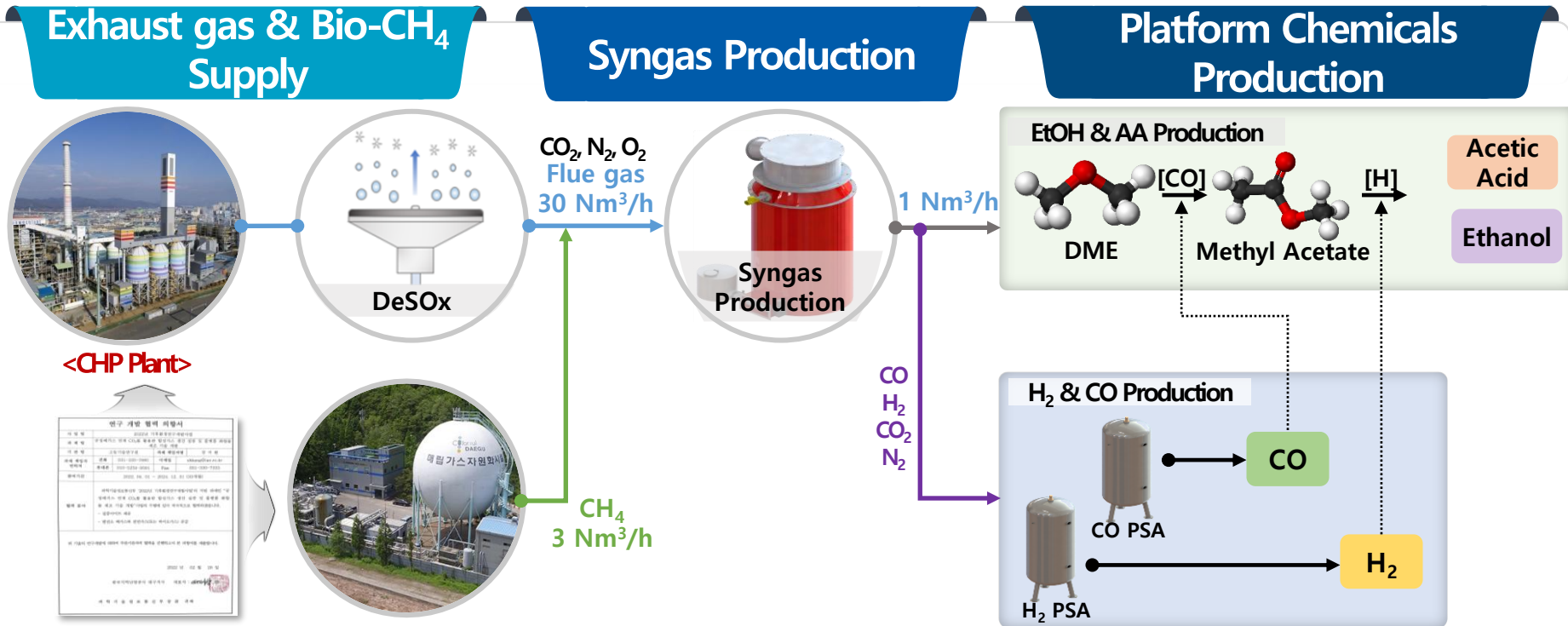


Experimental condition :

- 1) CH_4 & CO_2 feeding rate : $25.5 \text{ Nm}^3/\text{day}$
- 2) DRM temperature : $1,000^\circ\text{C}$
- 3) Product of Acetic Acid : $> 55 \text{ kg/day}$
- 4) A.A Reaction condi. : 190°C , 40 bar



Capture-free, Low-Concentration CO₂ Chemical Conversion (1)



Syngas Production Process Development

- Demonstration of a **30 t/y**-scale syngas production process.
- Development of a PDP for **10,000 t/y** commercial-scale process

Platform Chemical Production Technology Development

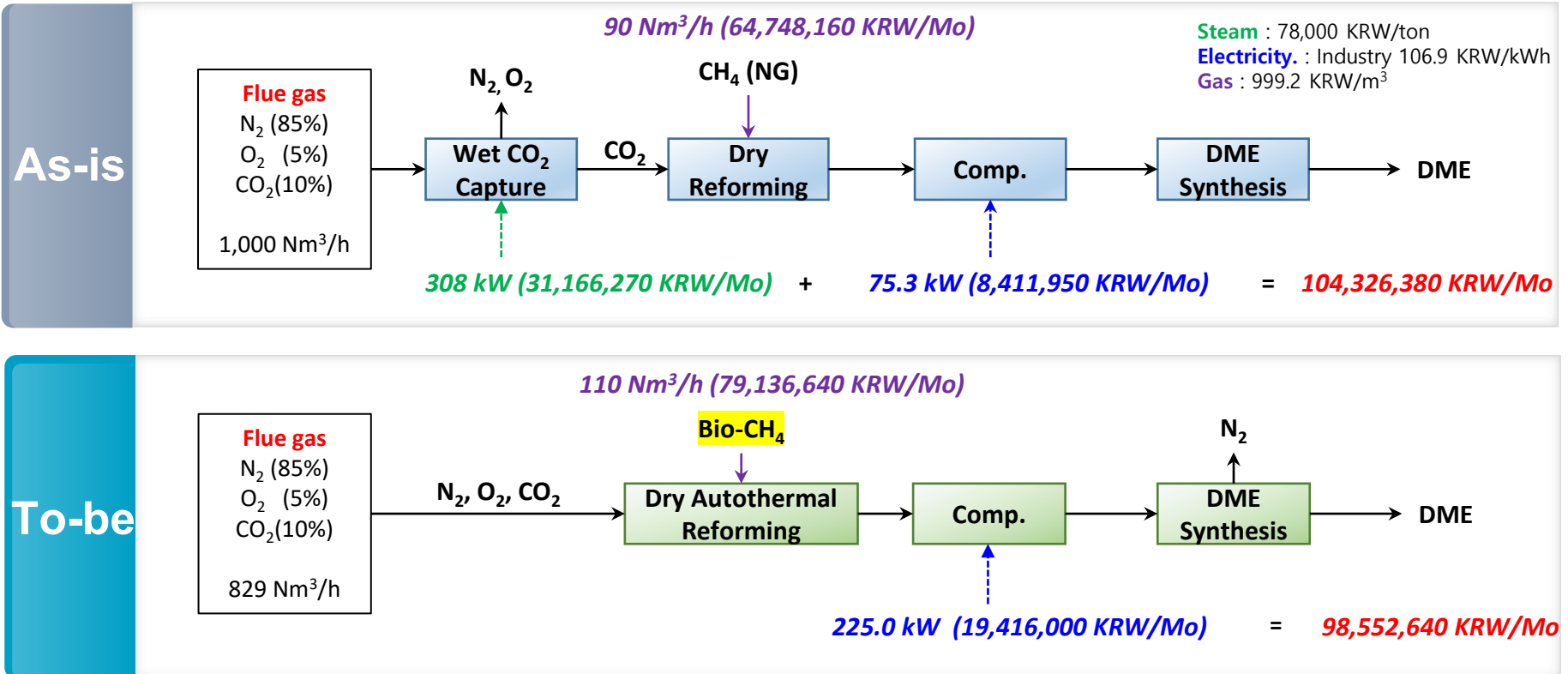
- **Construction & Operation of a platform chemical production process utilizing 1 Nm³/h of syngas**

Process Analysis & Efficiency Improvement

- **LCA and process evaluation**
- **Development of a Digital Twin Model**

Capture-free, Low-Concentration CO₂ Chemical Conversion (2)

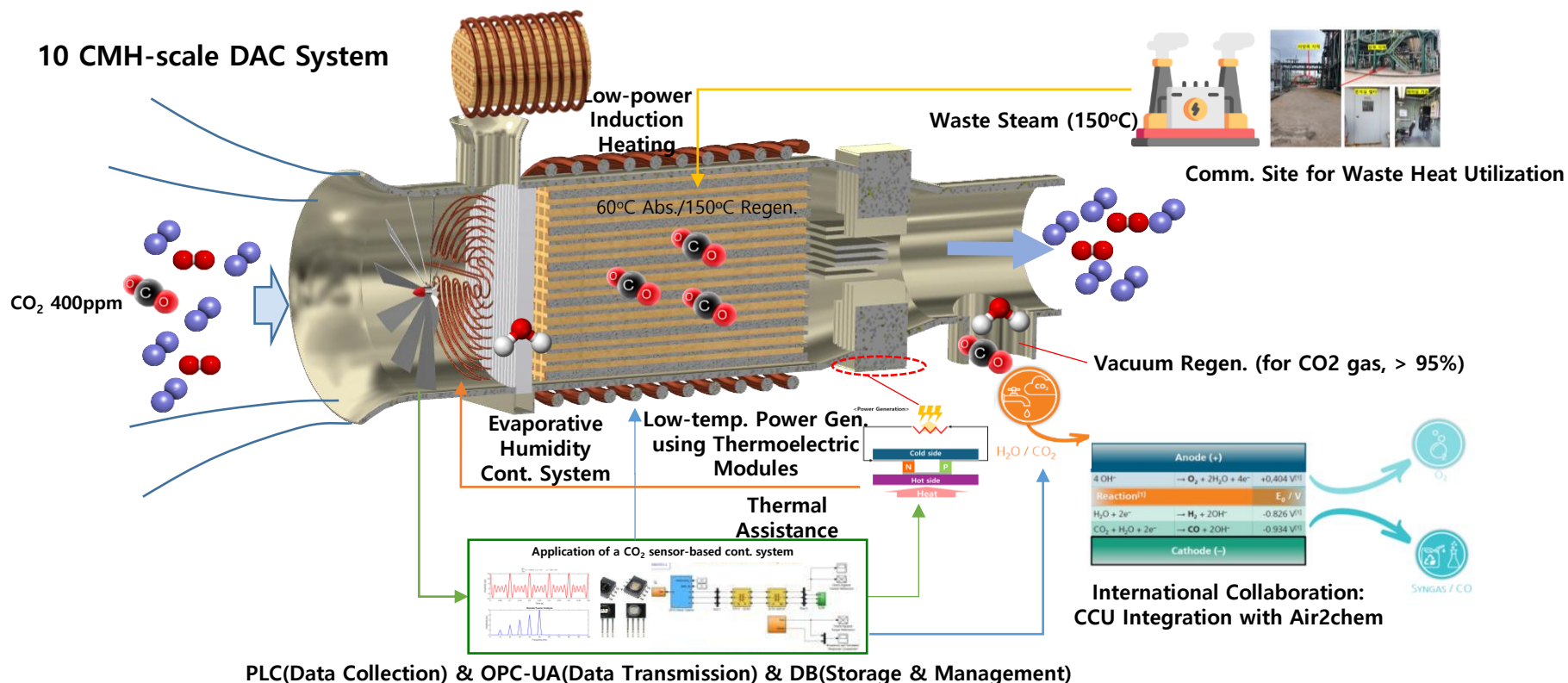
Comparison of Monthly Energy Costs for Wet CO₂ capture using Renewable Energy(LP Steam) and for Compression Including Nitrogen



Approximately **6% reduction** in operating costs compared to the conventional process.
Expected GHG reduction and cost saving through Bio-CH₄(renewable energy) supply.

DAC-CCU

10 CMH-scale DAC System



Comb. & Opti. of core tech. = Dry Sorbent + System Cont. + Regen. + Waste Heat + CCU Integration

Domestic DAC Regen. Energy Consumption (A) = Sorbent Regen. + System = **7.2GJ/CO₂ ton(A)**

Regen. Energy
★Reduction★

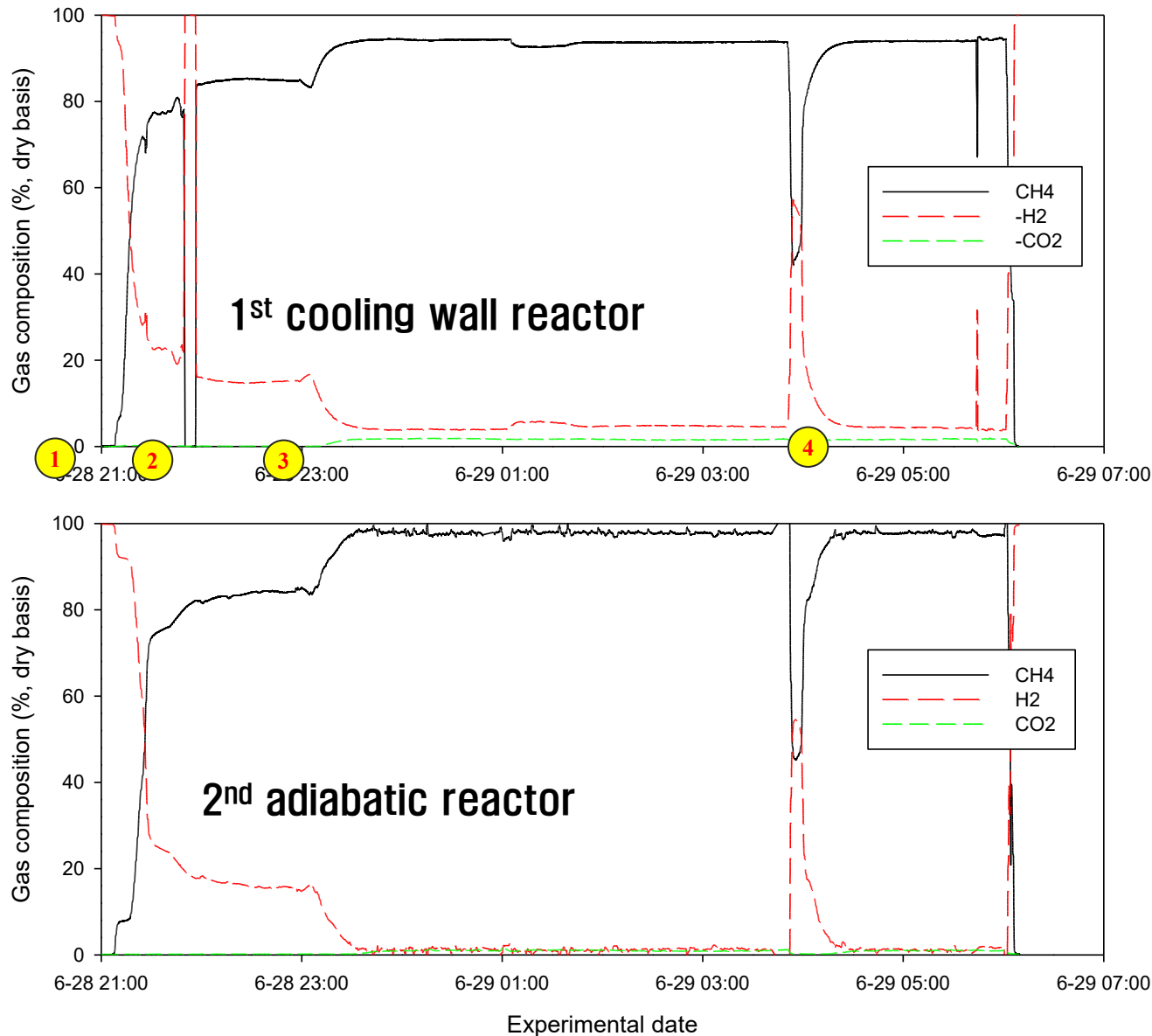
Waste-Heat Integration (Design of DAC body using tower and materials) Energy Savings = **1.53GJ/CO₂ ton (B)**

Low-power induction heating with thermoelectric materials, vacuum regen. System, Energy Savings = **1.7GJ /CO₂ ton(C)**

Regen. Energy of DAC system = **A-B-C = 3.97GJ/CO₂ ton**

About 45%Reduction

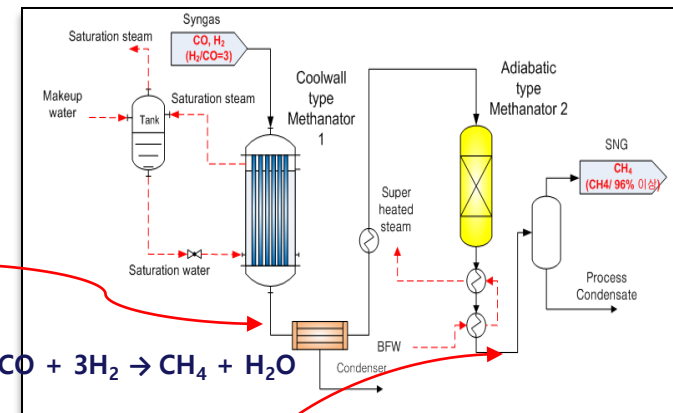
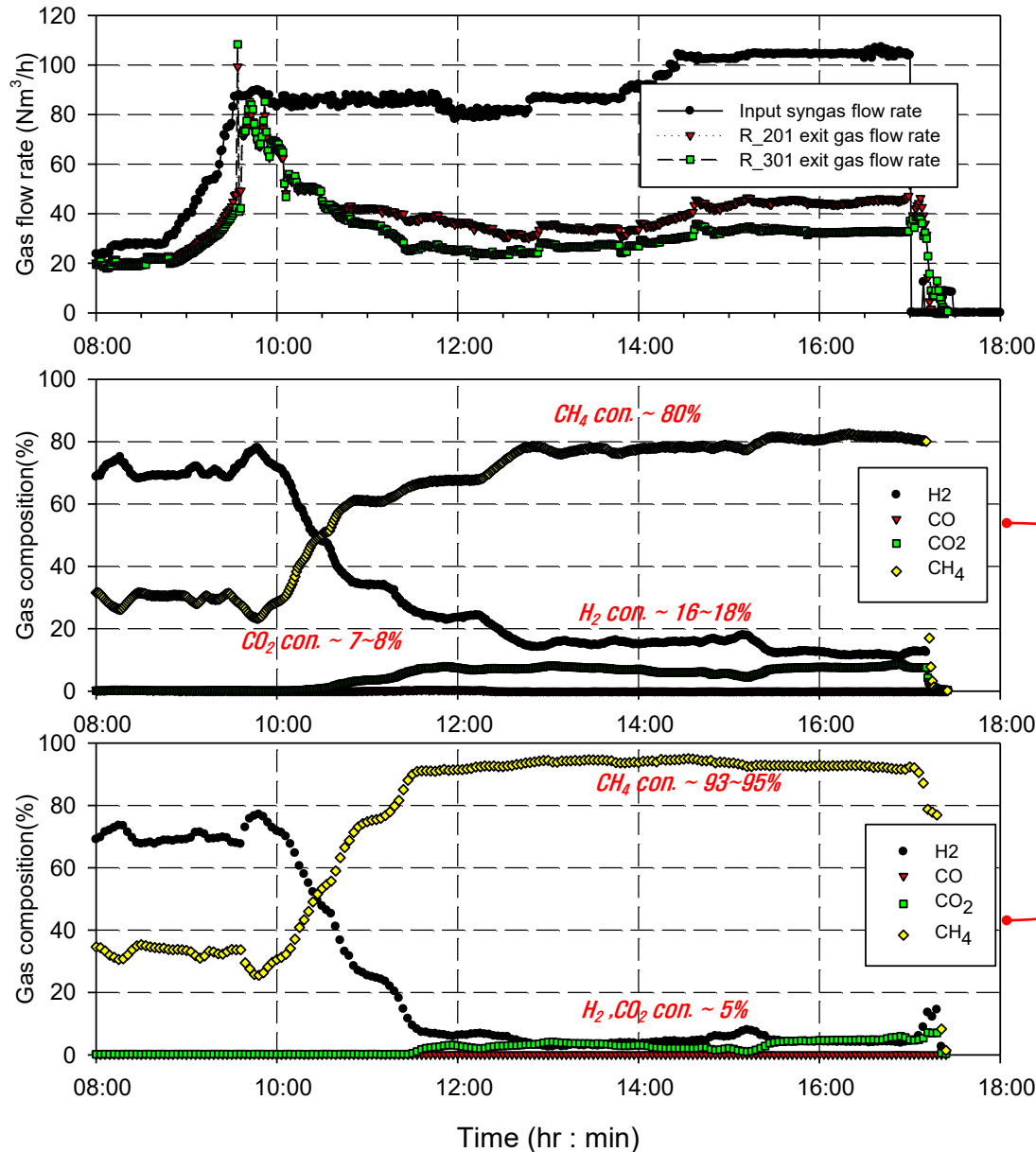
2 Nm³/h-scale Cooling Wall & Adiabatic Reactor System



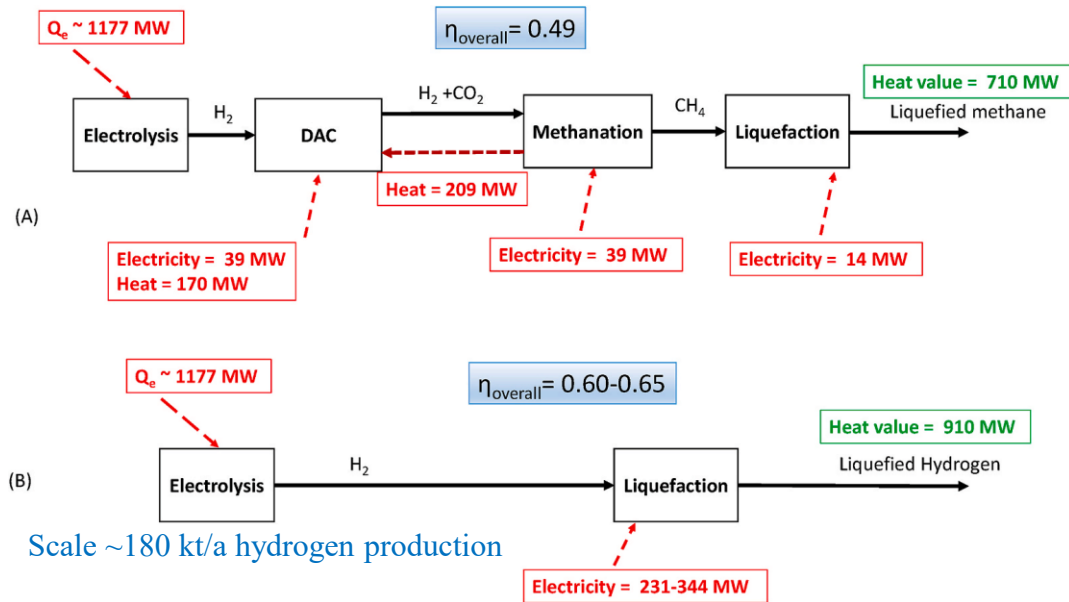
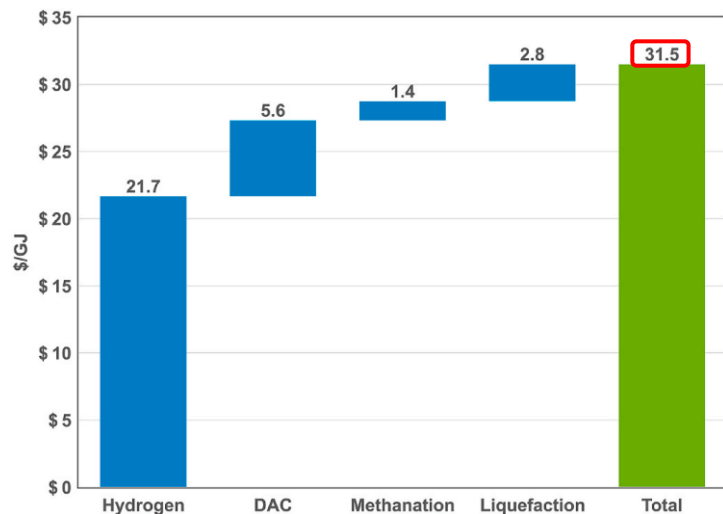
2 Nm³/h
H₂/CO=2.97

- ① Introduction of CO from a 20 bar hydrogen saturator (feeding rate: 1 Nm³/h)
- ② Analyzer malfunction occurred
- ③ Syngas (H₂/CO) flow increased to 2 Nm³/h.
- ④ During CO cylinder replacement, no CO was supplied.

100 Nm³/h-scale Cooling Wall & Adiabatic Reactor System

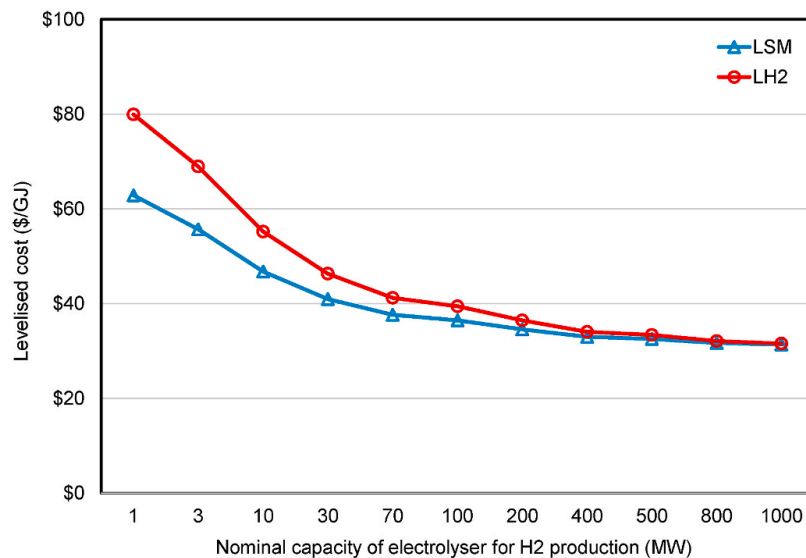


DAC-based LSM vs. LH₂



Scale ~180 kt/a hydrogen production

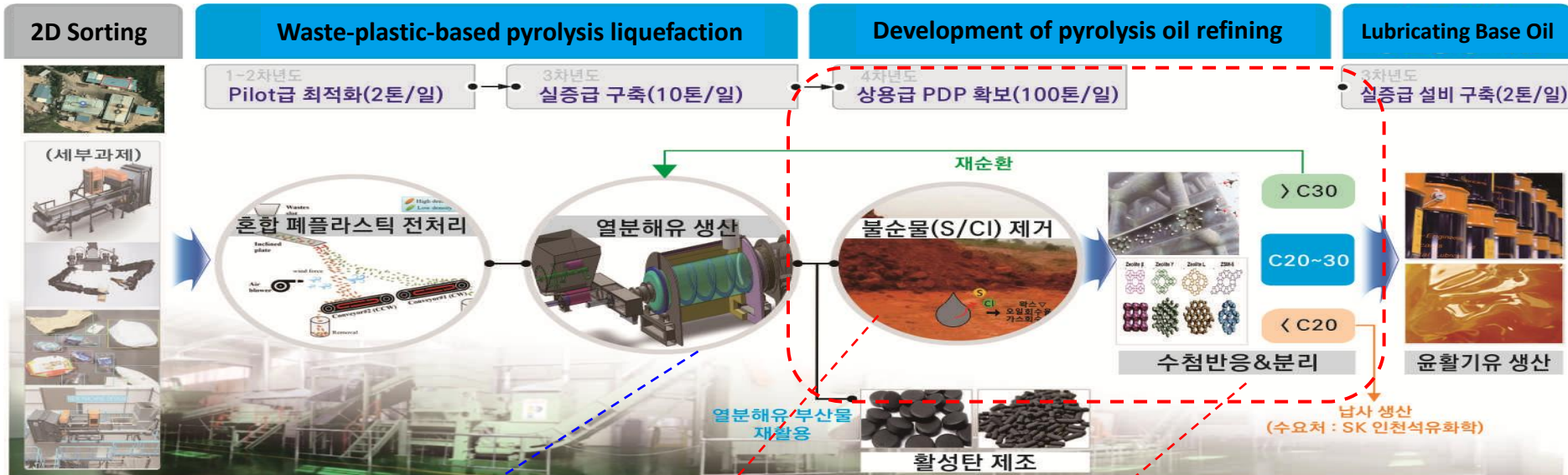
Liquefied synthetic methane : 360 kt/a
DAC



Costs of LSM and LH₂ per unit
Electrolysers efficiency ~80%

Source : Journal of Natural Gas Science and Engineering 94 (2021)

Recycled Fuel Oil Production using Waste Plastics



Removal of Impurities (Cl)

Provision of Adsorbent Design Conditions

Mass Production of Adsorbents and Design/Construction/Operation of the Adsorption

Development of Hydrotreating and Separation

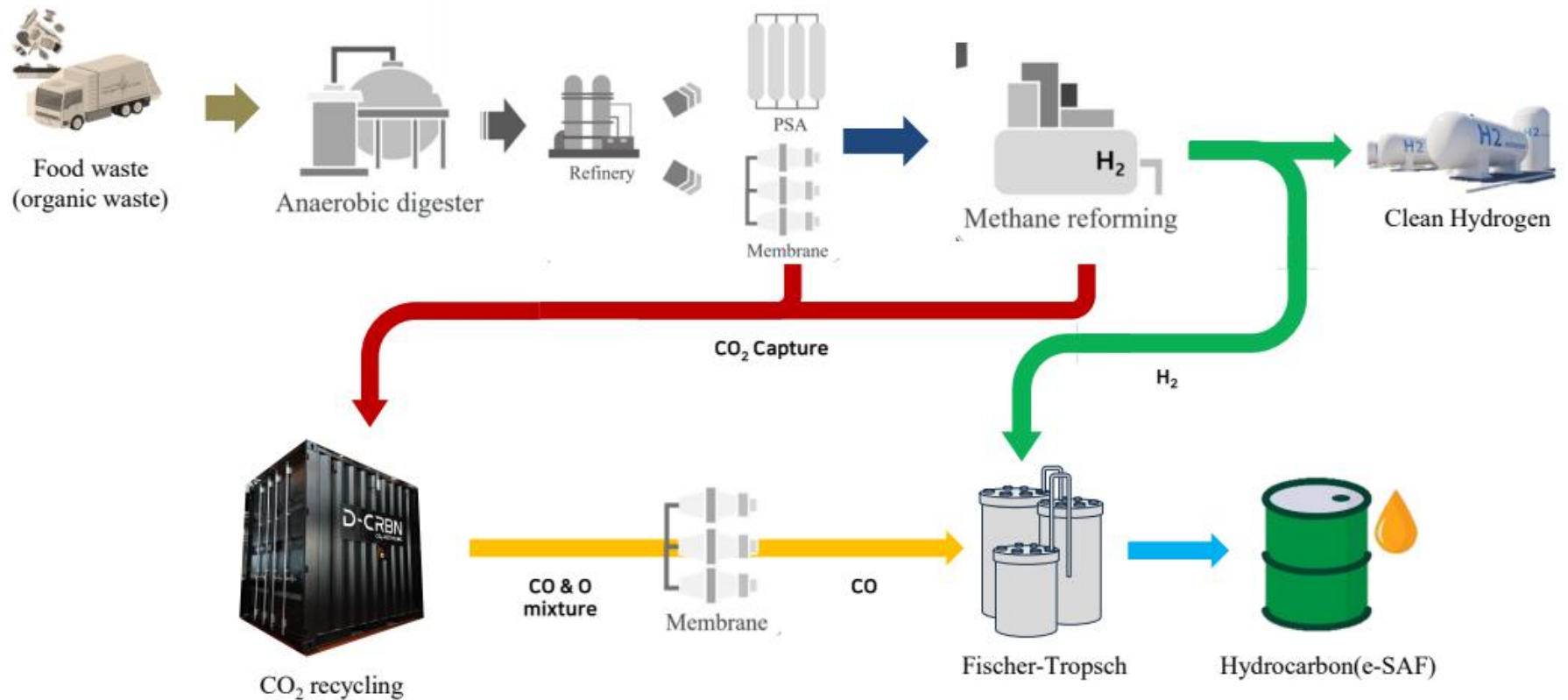
Provision of Catalyst Design

Mass Production of Catalyst and Design/Construction/Operation of the Hydrotreating



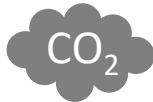
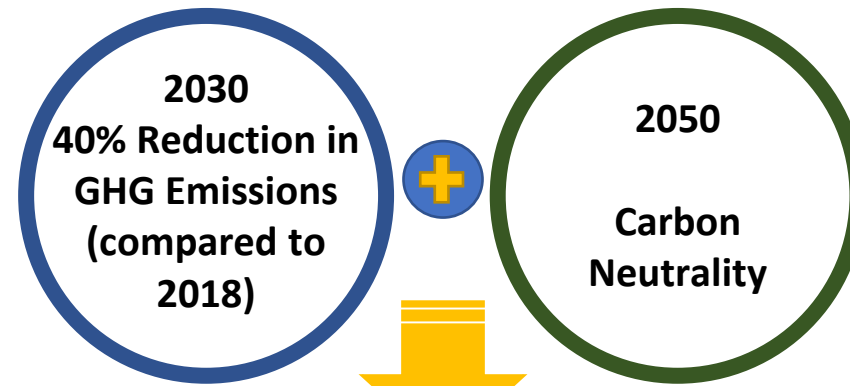
Control of T. P

Concept of Co-production Process for Clean H₂ & e-SAF from Biogas



Conclusion

Overcoming the Energy Crisis through Carbon Neutrality



Deployment of 'advanced energy transition technologies' is essential

Energy Transition
Technology

BECCU

Sustainable
Development

Towards an improved 'Quality of Life' 



Thank you

